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## Quiz 1

1. (10 pts. each) Determine whether the following molecules could undergo nucleophilic substitution via $\mathrm{S}_{\mathrm{N}} 1, \mathrm{~S}_{\mathrm{N}} 2$, both, or neither.
(s)
2. (20 pts. each) Predict the products for the following substitution reactions. Write complete balanced equations, and indicate the stereochemistry of the product(s) using wedge (- ) and dash ( $\cdot \cdots \cdot \boldsymbol{\prime \prime}$ III) bonds where appropriate.
$S_{N} 2$


3. (20 pts.) When a nucleophilic substitution occurs in the reaction below, the nucleophile adds to two places. Draw a mechanism that accounts for these products.
(Stereochemistry has been ignored in the reaction drawn below.

