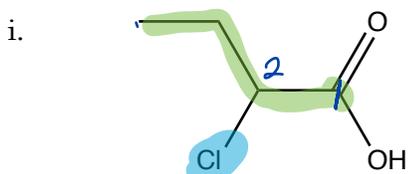
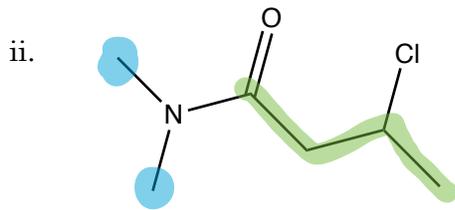


1. a. (2 pts. ea.) Identify the following compounds as acid chlorides, esters, carboxylic acids, or amides.  
 b. (4 pts. ea.) Provide IUPAC names for the following compounds.



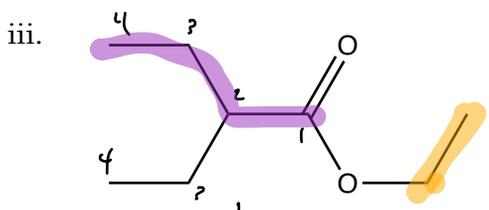
a. carboxylic acid

b. 2-chlorobutanoic acid



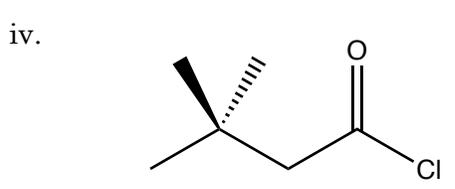
a. amide

b. N-chloro-N,N-dimethylbutanamide



a. ester

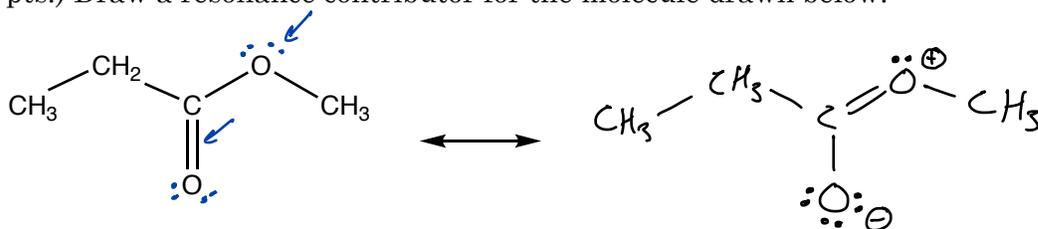
b. ethyl 2-ethylbutanoate



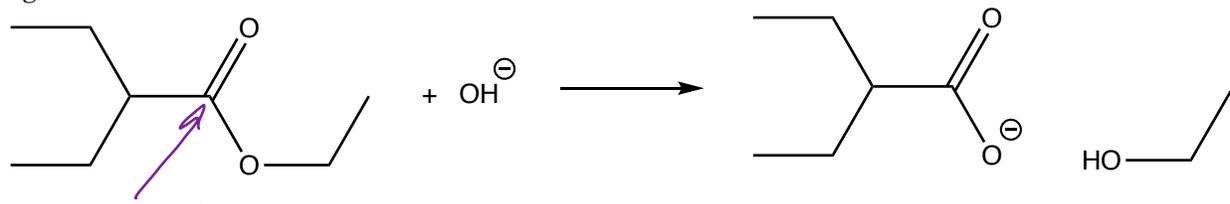
a. acid chloride

b. 3,3-dimethylbutanoyl chloride

2. (12 pts.) Draw a resonance contributor for the molecule drawn below.

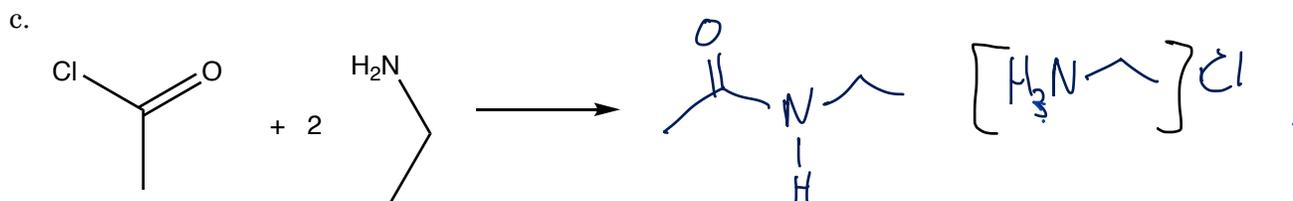
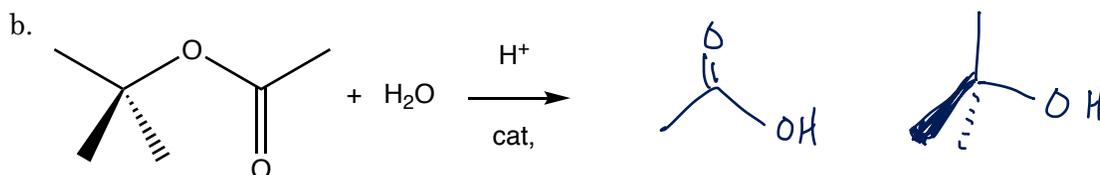
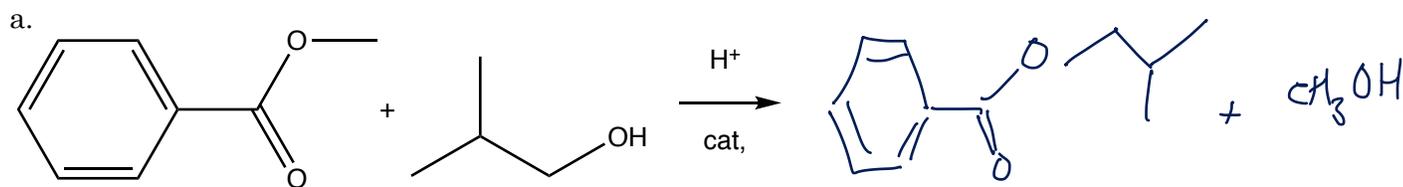


3. (12 pts) The reaction drawn below is not reversible. Briefly explain why the molecules on the right cannot react to form the molecules on the left.

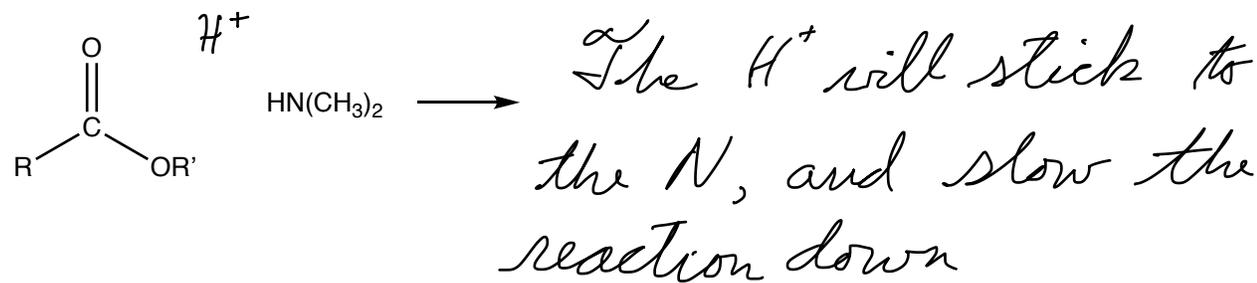


electrophilic  
 carboxylate ion is  $e^-$  rich and the extra  $e^-$  density repels the  $e^-$  rich nucleophile

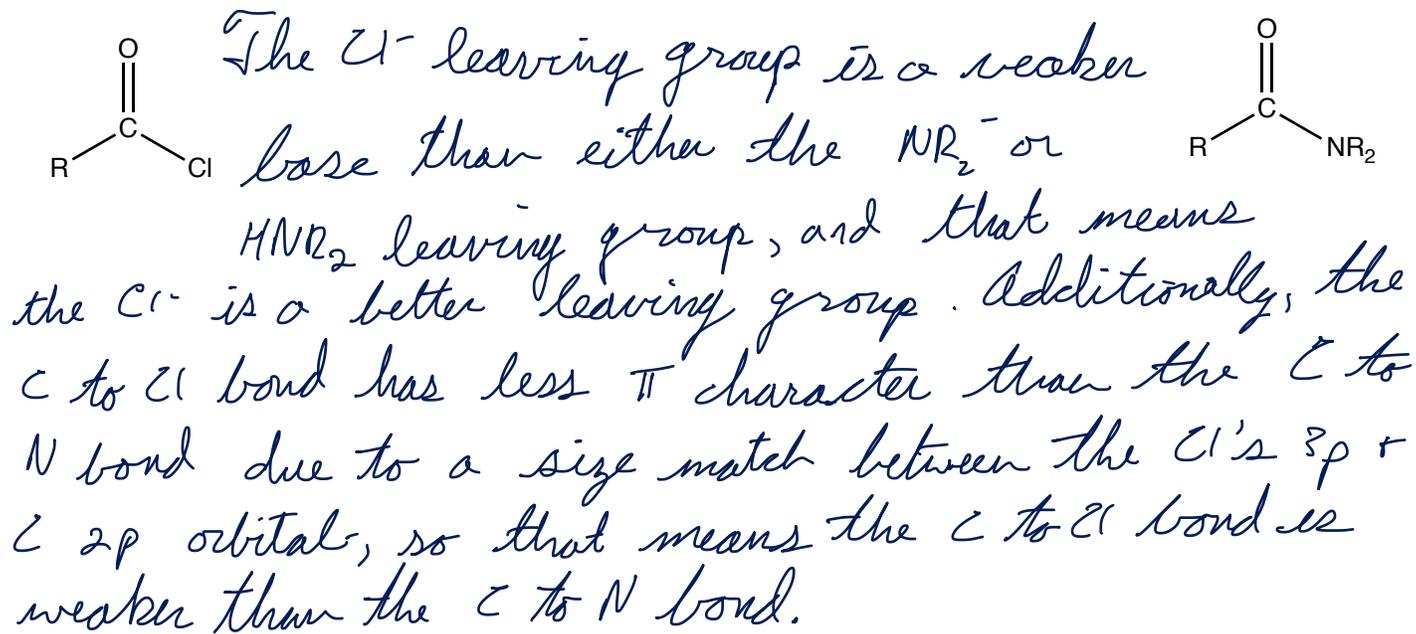
6. (6 pts. ea.) Predict the products for the following nucleophilic acyl substitution reactions.



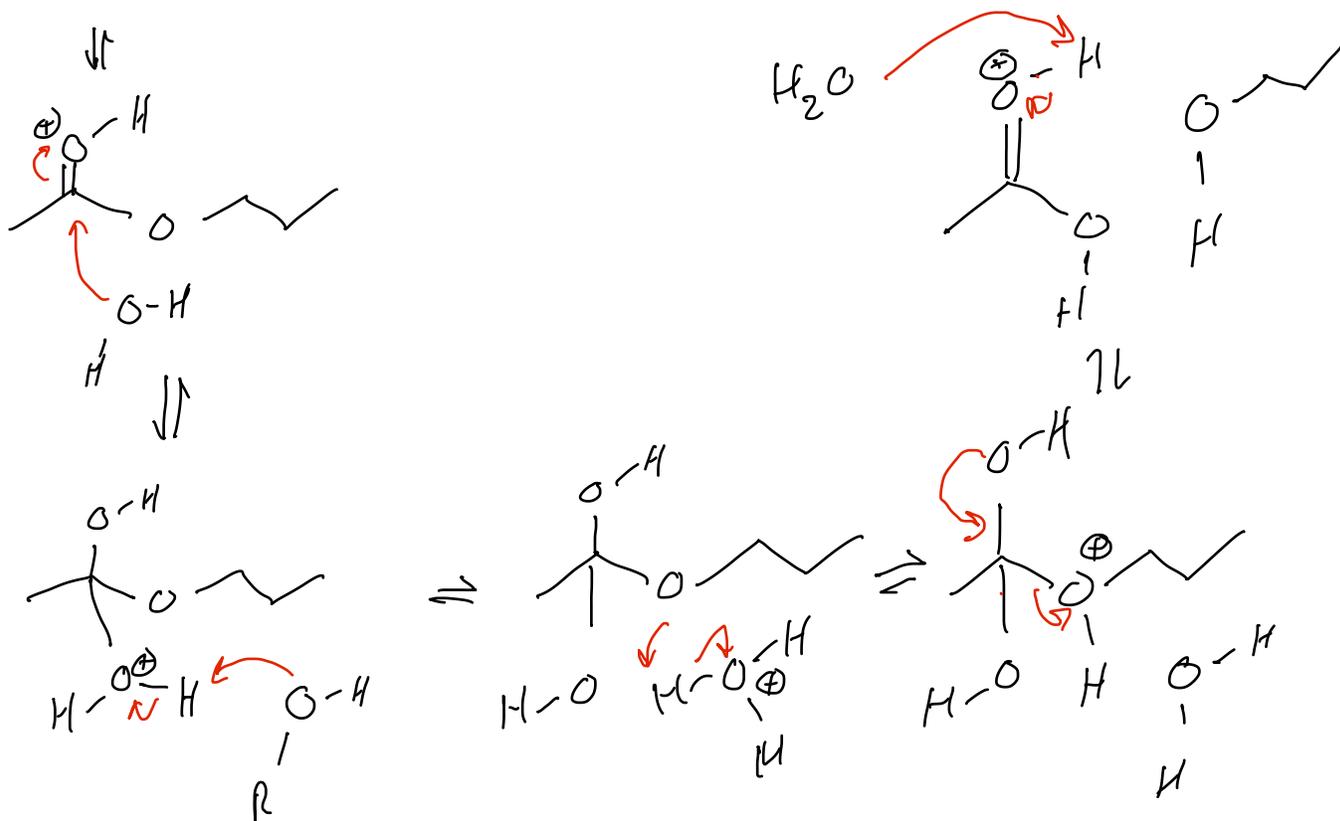
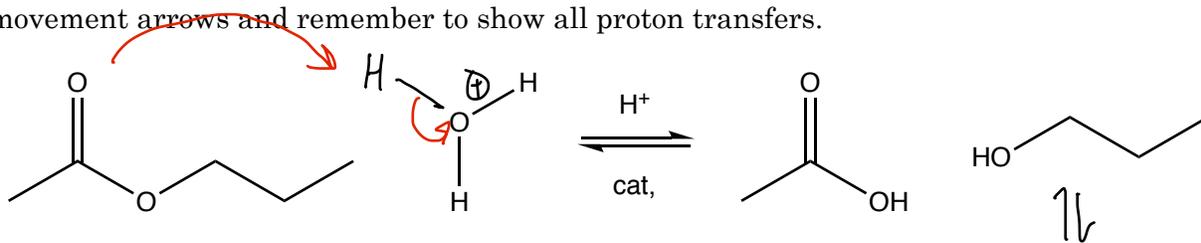
7. (12 pts.) Briefly explain why aminolysis of an ester cannot be catalyzed by the addition of acid.



8. (12 pts.) Briefly explain why acid chlorides are more reactive toward nucleophilic acyl substitution reaction than amides.



9. (12 pts.) Draw the mechanism for the hydrolysis of the following ester. Include electron movement arrows and remember to show all proton transfers.



1	<b>H</b> 1.0079																	2	<b>He</b> 4.0026		
3	4	<b>Li</b> 6.941	<b>Be</b> 9.012																	10	<b>Ne</b> 20.1797
11	12	<b>Na</b> 22.989	<b>Mg</b> 24.305																	18	<b>Ar</b> 39.948
19	20	<b>K</b> 39	<b>Ca</b> 40	<b>Sc</b> 21	<b>Ti</b> 22	<b>V</b> 23	<b>Cr</b> 24	<b>Mn</b> 25	<b>Fe</b> 26	<b>Co</b> 27	<b>Ni</b> 28	<b>Cu</b> 29	<b>Zn</b> 30	<b>Ga</b> 31	<b>Ge</b> 32	<b>As</b> 33	<b>Se</b> 34	<b>Br</b> 35	<b>Kr</b> 36		
37	38	<b>Cs</b> 56	<b>Sr</b> 57	<b>Y</b> 39	<b>Zr</b> 40	<b>Nb</b> 41	<b>Mo</b> 42	<b>Tc</b> 43	<b>Ru</b> 44	<b>Rh</b> 45	<b>Pd</b> 46	<b>Ag</b> 47	<b>Cd</b> 48	<b>In</b> 49	<b>Sn</b> 50	<b>Sb</b> 51	<b>Te</b> 52	<b>I</b> 53	<b>Xe</b> 54		
55	56	<b>Rb</b> 85	<b>Ba</b> 86	<b>La</b> 57	<b>Hf</b> 72	<b>Ta</b> 73	<b>W</b> 74	<b>Re</b> 75	<b>Os</b> 76	<b>Ir</b> 77	<b>Pt</b> 78	<b>Au</b> 79	<b>Hg</b> 80	<b>Tl</b> 81	<b>Pb</b> 82	<b>Bi</b> 83	<b>Po</b> 84	<b>At</b> 85	<b>Rn</b> 86		
87	88	<b>Fr</b> 87	<b>Ra</b> 88	<b>Ac</b> 89	<b>Rf</b> 104	<b>Db</b> 105	<b>Sg</b> 106	<b>Bh</b> 107	<b>Hs</b> 108	<b>Mt</b> 109	<b>110</b>	<b>111</b>	<b>112</b>	<b>114</b>	<b>116</b>					118	

58	<b>Ce</b> 58	<b>Pr</b> 59	<b>Nd</b> 60	<b>Pm</b> 61	<b>Sm</b> 62	<b>Eu</b> 63	<b>Gd</b> 64	<b>Tb</b> 65	<b>Dy</b> 66	<b>Ho</b> 67	<b>Er</b> 68	<b>Tm</b> 69	<b>Yb</b> 70	<b>Lu</b> 71
90	<b>Th</b> 90	<b>Pa</b> 91	<b>U</b> 92	<b>Np</b> 93	<b>Pu</b> 94	<b>Am</b> 95	<b>Cm</b> 96	<b>Bk</b> 97	<b>Cf</b> 98	<b>Es</b> 99	<b>Fm</b> 100	<b>Md</b> 101	<b>No</b> 102	<b>Lr</b> 103