

Today

Practice Interpreting Spectral Data

Introduction to Carbonyls: Nomenclature and Resonance

Sections 15.1 – 15.3

Next Class

Reactions of Carboxylic Acids and Carboxylic Acid Derivatives

Sections 15.4 -15.9

Second Class from Today

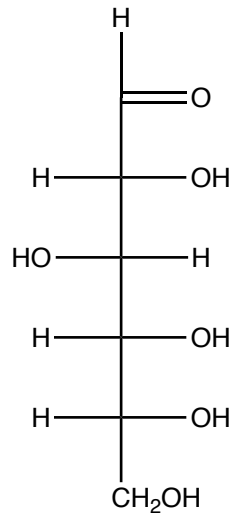
Reaction of Amides , Nitriles, and Acid Anhydrides

Sections 15.10 – 15.16

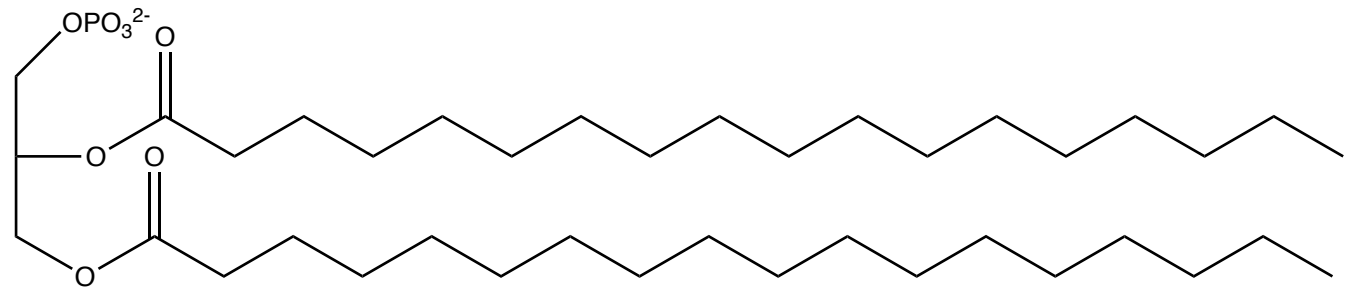
Third Class from Today

Test on Chap 10.1 – 10.4, Chap 13 (MS and IR), Chap 14 (NMR)

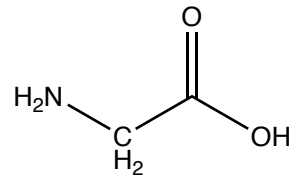
Why Carbonyls, Carboxylic Acids, and Carboxylic Acid Derivatives?



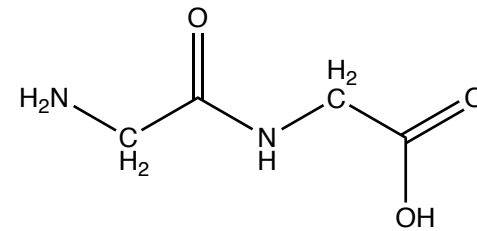
sugar



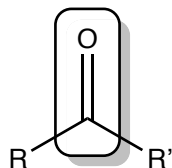
Phospholipid



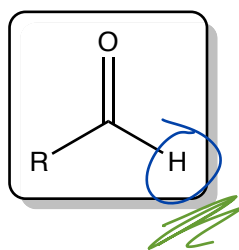
Amino Acid



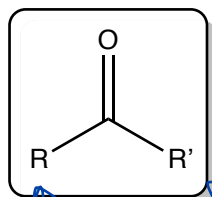
dipeptide



carbonyl means $C=O$ carbonyl C
 atoms are electrophilic + they react with
 nucleophiles



aldehyde
 aldehydes can
 do chemistry
 that ketones can't

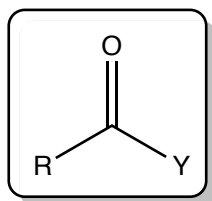


R, R' ≠ H

ketone

carbon atoms

these two classes of molecules
 do similar chemistry with
 nucleophiles

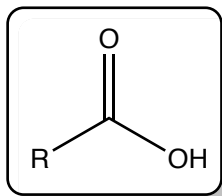


$Y \neq C, H$

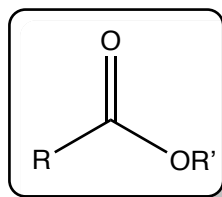
Not C or H

something
eneq next
to carbonyl

also react
with nucleophiles
but can do
chemistry that
aldehydes and
ketones don't
do because
of the
presence of
the eneg
atom

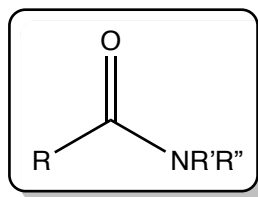


carboxylic acid

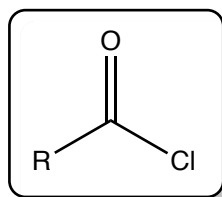


$R' \neq H$

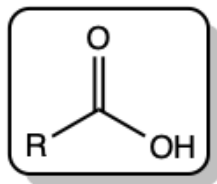
ester



amide



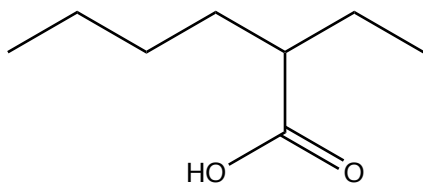
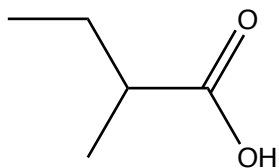
acid chlorides



Longest chain that starts with the functional group

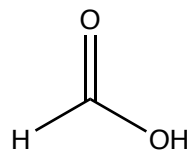
Remove the “e” and add “oic acid”

Place substituents in front of the name of the acid and number starting at the carbonyl carbon

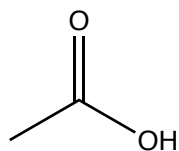


“Common Names”

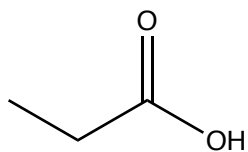
Section 15.1



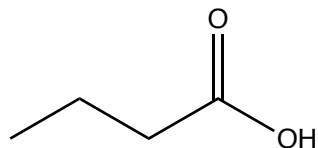
formic acid



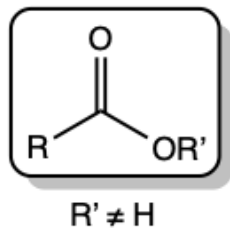
acetic acid



propionic acid

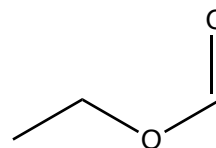
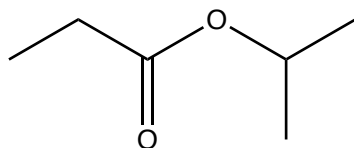


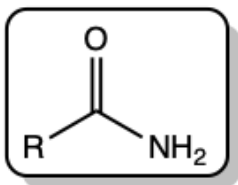
butyric acid



“OR group name” “carbonyl group name”

1. Group bonded to O
 - a. named as though it is an alkyl substituent; that is, longest chain starting at the O, drop the “e” and add “yl”
 - b. add any substituents to the beginning of the alkyl group’s name
2. Name the carbonyl group
 - a. the longest chain that starts with the C=O
 - b. drop the “e” and add “oate”
3. name substituents by adding them to the beginning of the carbonyl group name



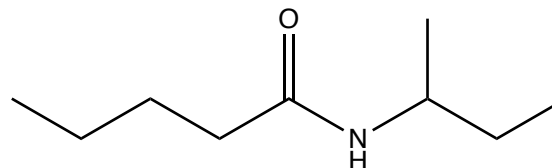
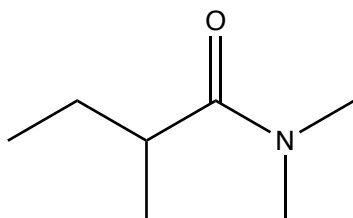


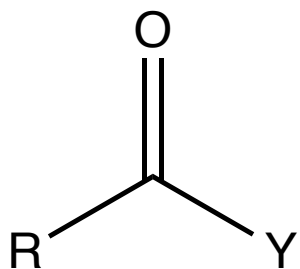
“N-substituent” “carbonyl group name”

1. Groups bonded N are named as N-alkyl substituents
 - a. longest chain starting at the N, drop the “e” and add “yl”
 - b. substituents on alkyl group
 - i. place in parenthesis at the beginning of the alkyl group’s name
 - ii. C atom connected to N is C-1 on the N-substituent

2. Name the carbonyl group
 - a. the longest chain that starts with the C=O
 - b. drop the “e” and add “amide”

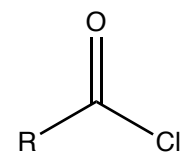
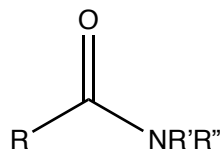
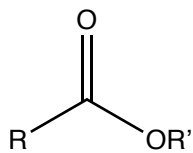
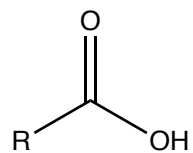
3. name other substituents by adding them to the beginning of the carbonyl group name





Resonance in Carboxylic Acids and Acid Derivatives

Section 15.2



Nucleophilic Acyl Substitution

