Today

Next Class

Sections 4.3 - 4.8 Chirality Sections 4.9-4.14 Optical activity and compounds with more than one center of chirality

Rework test 1 by Oct 28. Answers questions that didn't receive full credit on a separate piece of paper. I do not need your test back

Isomers



Look down



## What makes your feet chiral?

A chiral object has a non-superposable mirror image

An achiral object has a superposable mirror image

Superposable mean that when you superimpose the two objects everything lines up.



## What makes your feet chiral?

A chiral object has a non-superposable mirror image

Superposable mean that when you superimpose the two objects everything lines up.



A chiral object lacks an internal mirror plane\*



\*Technically it's an improper axis of rotation, but a mirror plane is an  $S_1$  and a center of inversion is an  $S_2$ 







₽-glucose 6¢ per gram

L-glucose \$103 per gram





L-alanine

D-alanine

Practice Recognizing centers of chirality









Section 4.4, 4.13

Section 4.8 clockwise Determining Configuration (R vs S) Same rules as Z+E, except 1<sup>st</sup>, 2<sup>nd</sup>, 3<sup>rd</sup>, + 4<sup>th</sup> place Assign priorities to groups connected to chirality center

Point lowest priority group away

Draw a circle from 1<sup>st</sup> to 2<sup>nd</sup> to 3<sup>rd</sup> priority groups

**Clockwise** circle is **R** configuration

**Counter Clockwise** circle is **S** configuration

## Determining Configuration (R vs S)

















- 1. Draw a tetrahedral C atom
- 2. Assign priorities to the groups
- 3. Place the lowest priority group so that it points away
- 4. Draw in priority groups 1 through 3 in the correct (clockwise or counterclockwise) orientation.

1. Draw the molecule

2. Assign priorities and check if the correct configuration is drawn

- 3. a. If correct, celebrate, you're done
- 3. b. If incorrect version is drawn, redraw molecule switching the positions of 2 (and only two) substituents.

R-2-chloropentane

(2S,3S)-2-bromo-3-chloropentane