

(16) **Today**

Chap 12 & 13: Practice Identifying Unknowns

Chap 21.1: Nomenclature

Chap 21.2: Nucleophilic Acyl Substitution

Next Class (17)

Chap 21.2: Nucleophilic Acyl Substitution

Chap 21.3: Reactions of Carboxylic Acids

(18) **Second Class from Today**

Chap 21.3: Reactions of Carboxylic Acids

Chap 21.4: Reactions of Acid Halides

Chap 21.6: Reactions of Esters

Third Class from Today (19)

Chap 21.4: Reactions of Acid Halides

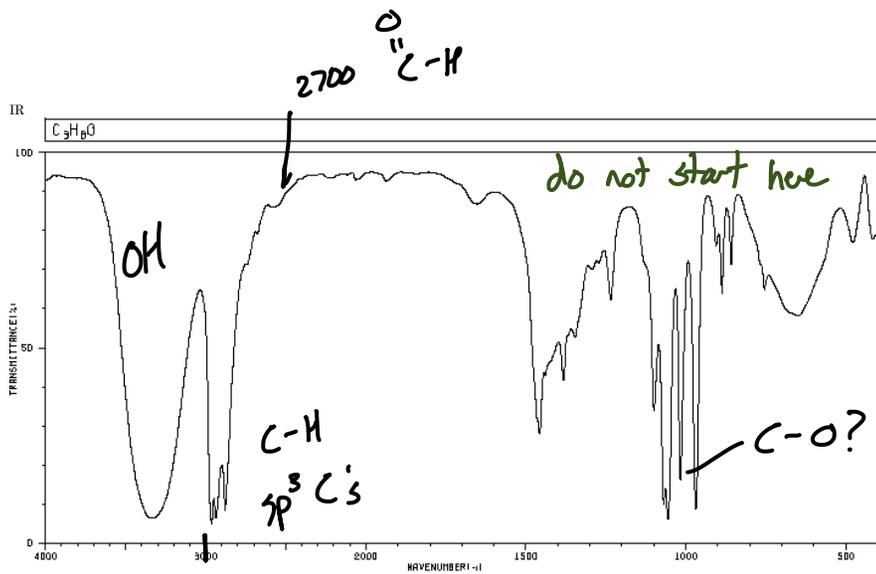
Chap 21.6: Reactions of Esters

Chap 21.7: Reactions of Amides

Please hand in reworked test 1.

Please remember, this assignment does not add points back onto your test score. This is a separate assignment that is worth 5% of your overall grade.

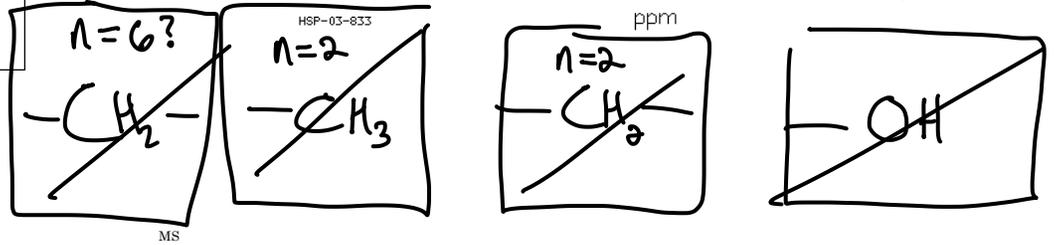
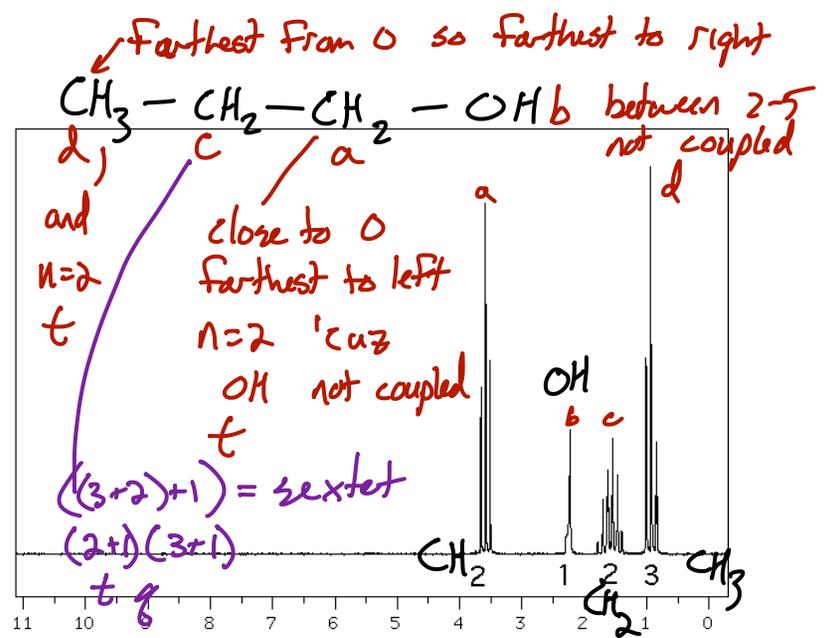
I do not need your test back, please just hand in the reworked answers.



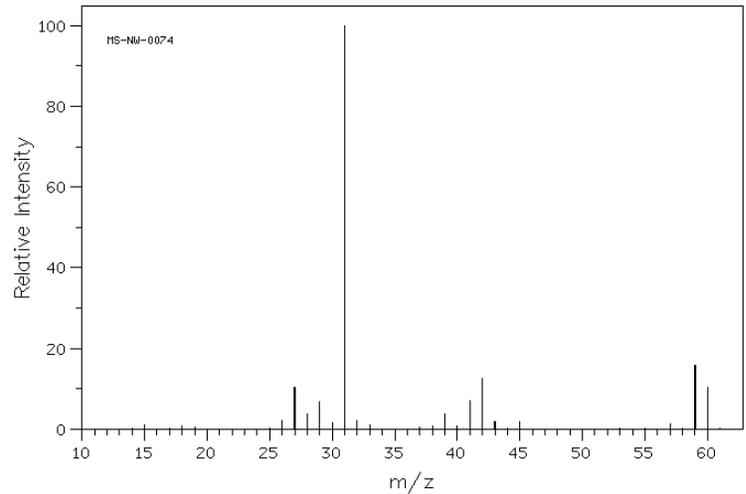
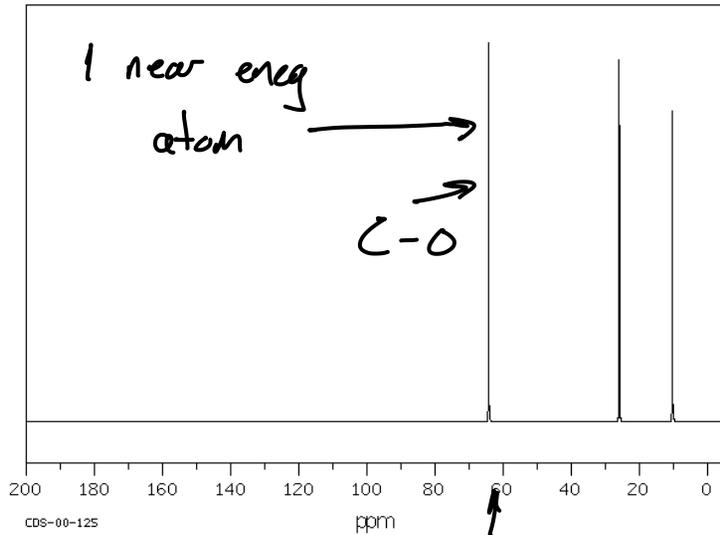
| | | | | | |
|------|----|------|----|------|----|
| 3333 | 6 | 1383 | 38 | 1017 | 16 |
| 2953 | 4 | 1346 | 50 | 969 | 8 |
| 2936 | 6 | 1293 | 66 | 905 | 72 |
| 2879 | 8 | 1235 | 60 | 898 | 62 |
| 1656 | 84 | 1100 | 32 | 858 | 68 |
| 1651 | 84 | 1063 | 8 | 755 | 82 |
| 1456 | 26 | 1056 | 6 | 479 | 74 |

1H NMR

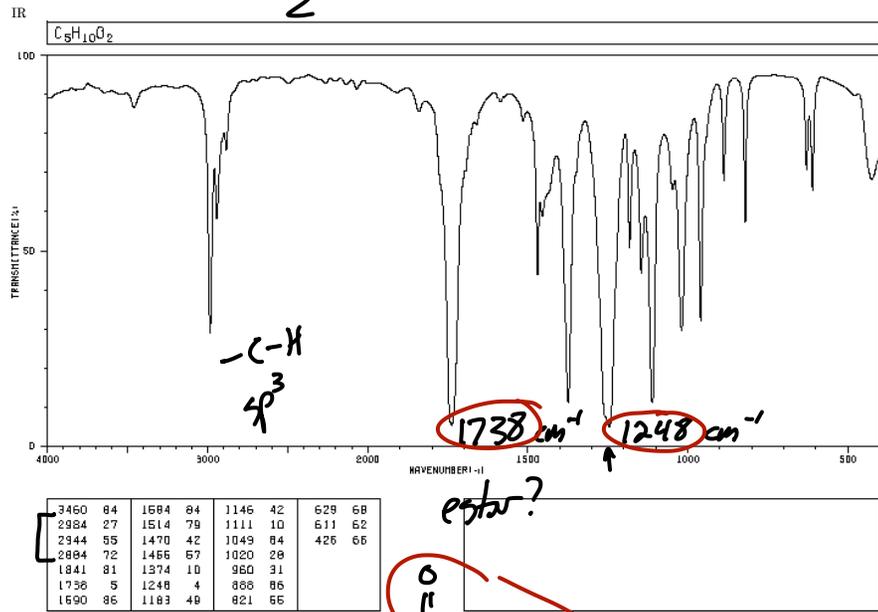
| ppm | Int. |
|-------|------|
| 3.658 | 431 |
| 3.585 | 906 |
| 3.511 | 501 |
| 2.225 | 322 |
| 1.717 | 49 |
| 1.699 | 143 |
| 1.629 | 216 |
| 1.544 | 300 |
| 1.467 | 204 |
| 1.46 | 103 |
| 1.398 | 60 |
| 1.009 | 508 |
| 0.932 | 1000 |
| 0.846 | 289 |



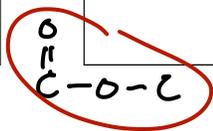
^{13}C NMR 3 chemically distinct C atoms



$$\frac{2 \cdot 5 + 2 - 10}{2} = 1$$

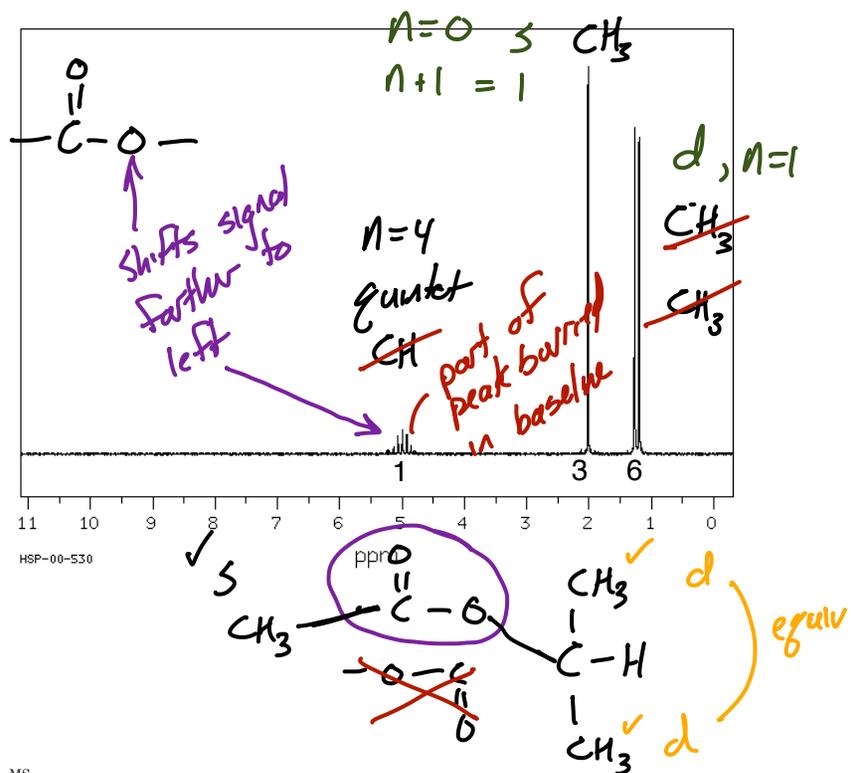


| | | | | | | | |
|------|----|------|----|------|----|-----|----|
| 3460 | 84 | 1584 | 84 | 1146 | 42 | 629 | 68 |
| 2984 | 27 | 1514 | 79 | 1111 | 10 | 511 | 62 |
| 2944 | 55 | 1470 | 42 | 1049 | 84 | 426 | 66 |
| 2884 | 72 | 1466 | 67 | 1020 | 29 | | |
| 1841 | 81 | 1374 | 10 | 960 | 31 | | |
| 1736 | 5 | 1248 | 4 | 868 | 86 | | |
| 1690 | 86 | 1183 | 48 | 821 | 66 | | |



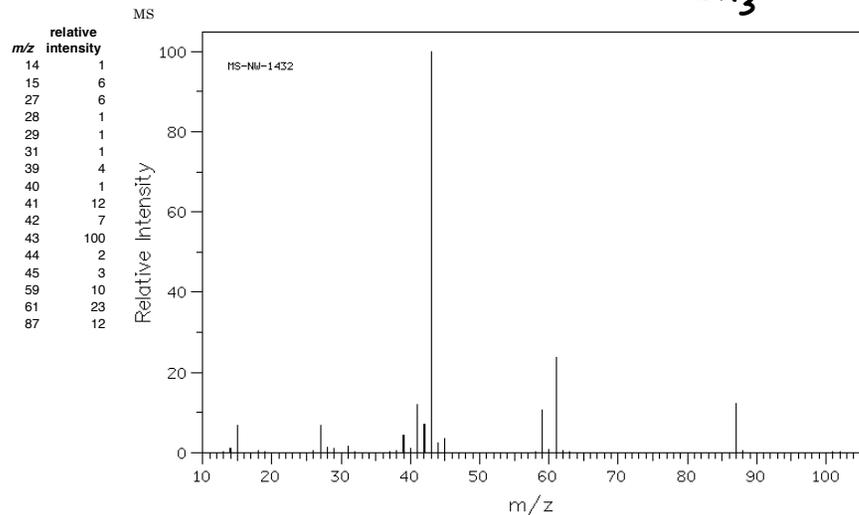
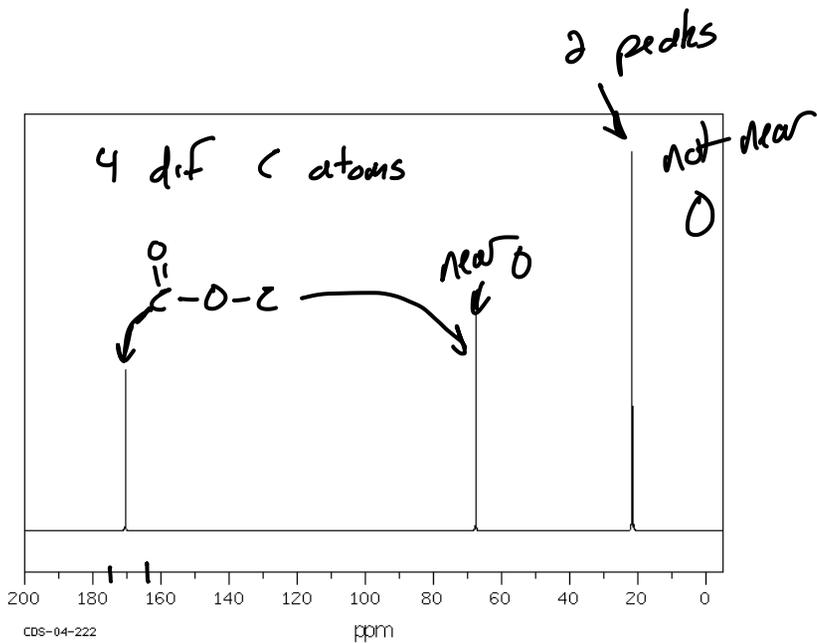
¹H NMR

| ppm | Int. |
|-------|------|
| 5.134 | 17 |
| 5.063 | 47 |
| 4.993 | 64 |
| 4.923 | 50 |
| 4.853 | 19 |
| 2.017 | 1000 |
| 1.268 | 842 |
| 1.198 | 818 |



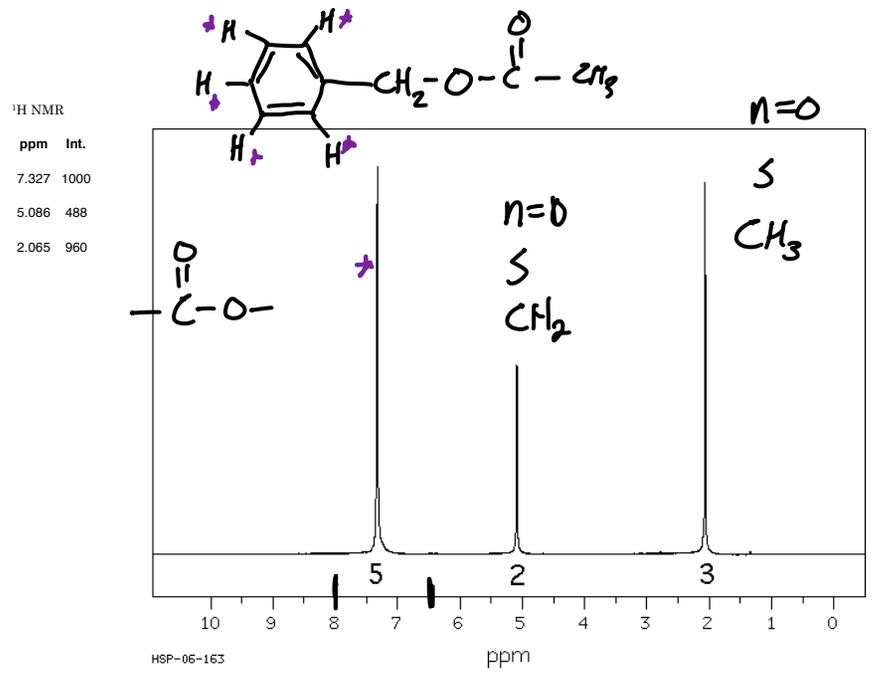
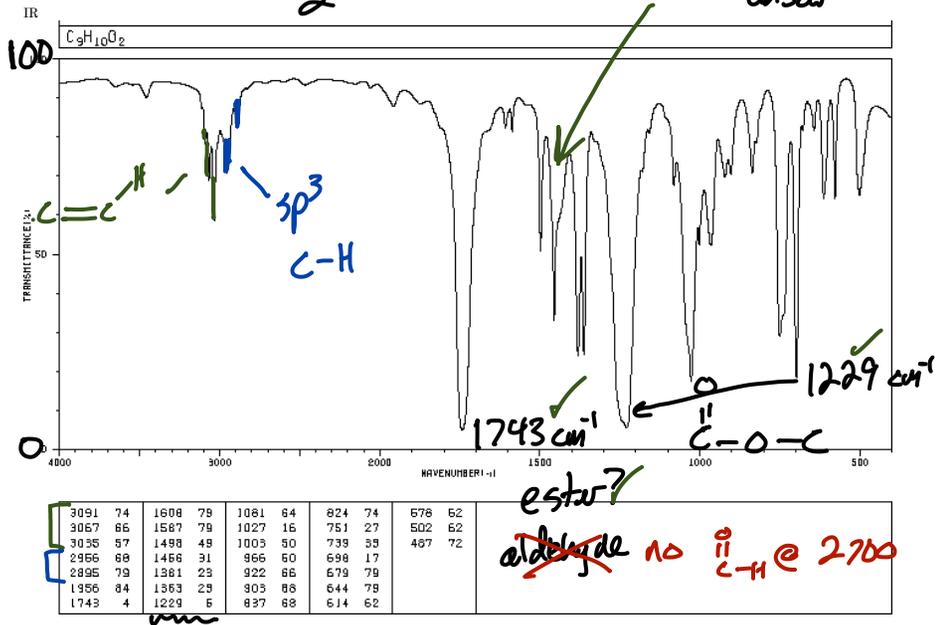
¹³C{¹H} NMR

| ppm | Int. |
|--------|------|
| 170.42 | 425 |
| 67.52 | 570 |
| 21.79 | 1000 |
| 21.34 | 325 |

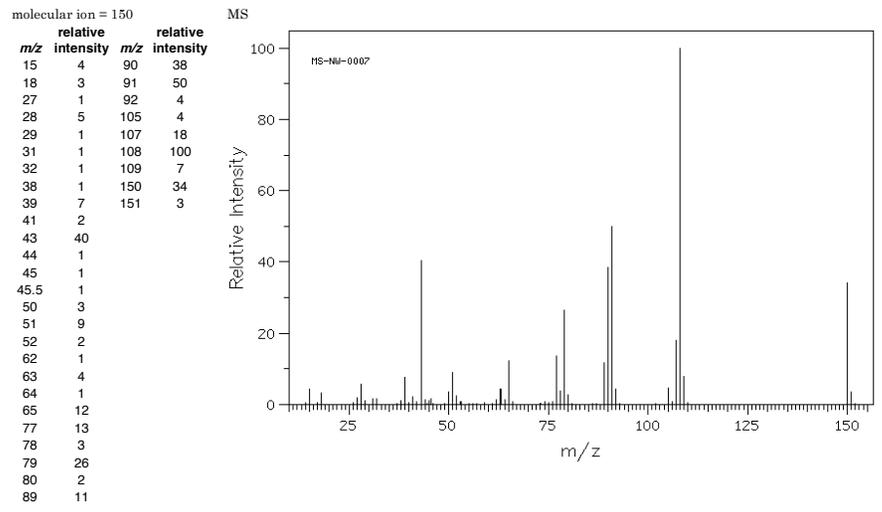
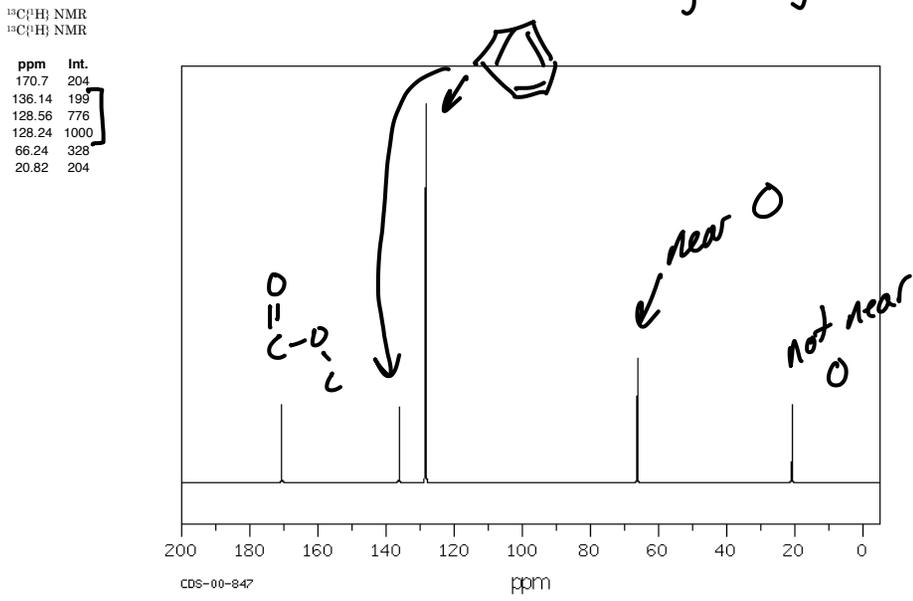


$$\frac{(9 \times 2 + 2) - 10}{2} = 5$$

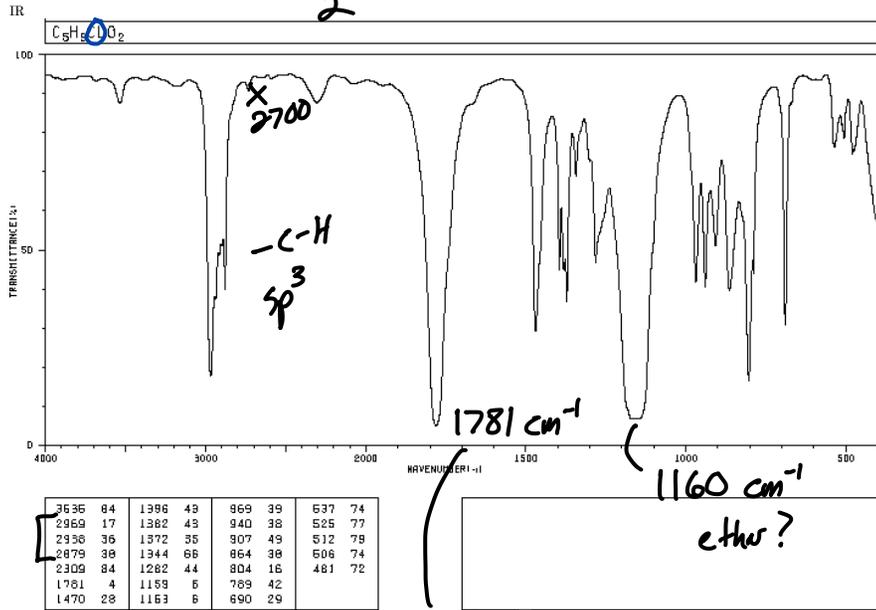
← 4° of unsat



6 dif C atoms ... symmetry



$$\frac{(5.2 + 2 - 0) - 9}{2} = 1$$

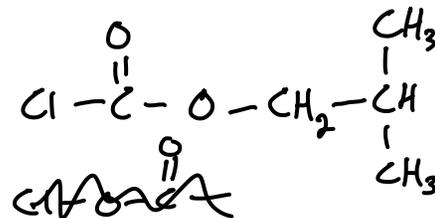
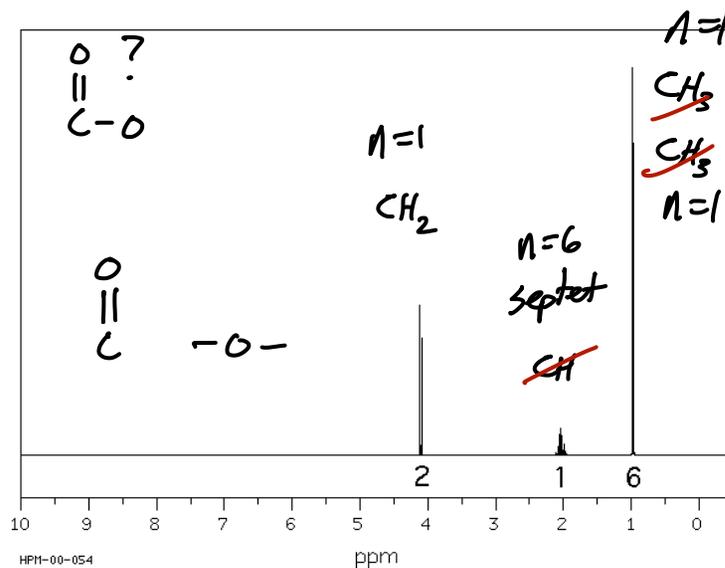


ester? 1250-1200



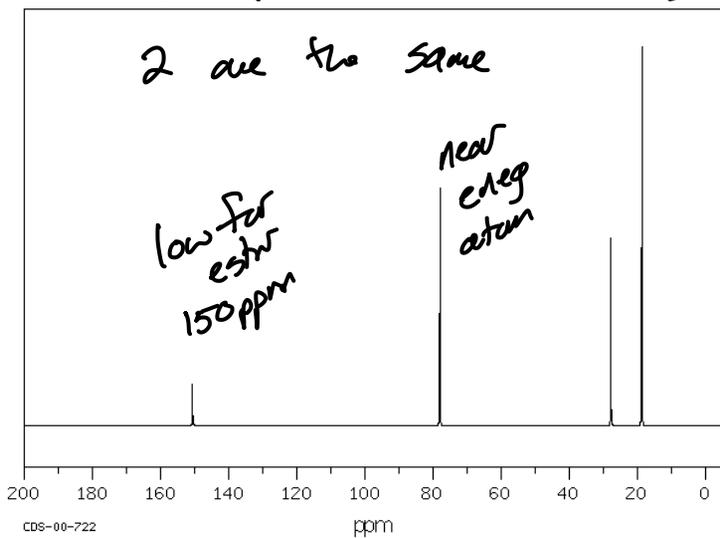
¹H NMR

| ppm | Int. |
|-------|------|
| 4.118 | 335 |
| 4.096 | 343 |
| 2.113 | 10 |
| 2.091 | 32 |
| 2.069 | 62 |
| 2.046 | 79 |
| 2.024 | 65 |
| 2.002 | 35 |
| 1.98 | 11 |
| 0.992 | 1000 |
| 0.97 | 957 |



¹³C{¹H} NMR

4 dif C atom CH₂CH₃CH
2 are the same



molecular ion = 136

| m/z | relative intensity |
|-----|--------------------|
| 15 | 1 |
| 26 | 1 |
| 27 | 21 |
| 28 | 4 |
| 29 | 21 |
| 31 | 4 |
| 36 | 1 |
| 38 | 1 |
| 39 | 15 |
| 40 | 3 |
| 41 | 71 |
| 42 | 16 |
| 43 | 100 |
| 44 | 6 |
| 53 | 1 |
| 55 | 8 |
| 56 | 76 |
| 57 | 33 |
| 58 | 1 |
| 59 | 1 |
| 63 | 37 |
| 65 | 12 |
| 93 | 1 |
| 94 | 8 |
| 95 | 4 |
| 96 | 2 |
| 97 | 1 |

